

## RESEARCH PAPER

# Histamine H<sub>3</sub>-receptor agonists and imidazole-based H<sub>3</sub>-receptor antagonists can be thermodynamically discriminated

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**Background and purpose:** Studies suggest that measurement of thermodynamic parameters can allow discrimination of agonists and antagonists. Here we investigate whether agonists and antagonists can be thermodynamically discriminated at histamine H<sub>3</sub> receptors.

**Experimental approach:** The pK<sub>L</sub> of the antagonist radioligand, [<sup>3</sup>H]-clobenpropit, in guinea-pig cortex membranes was estimated at 4, 12, 21 and 30°C in 20mM HEPES-NaOH buffer (buffer A), or buffer A containing 300mM CaCl<sub>2</sub>, (buffer A<sub>Ca</sub>). pK<sub>i</sub>' values for ligands with varying intrinsic activity were determined in buffer A and A<sub>Ca</sub> at 4, 12, 21 and 30°C.

**Key results:** In buffer A, the pK<sub>L</sub> of [<sup>3</sup>H]-clobenpropit increased with decreasing temperature while it did not change in buffer A<sub>Ca</sub>. The B<sub>max</sub> was not affected by temperature or buffer and n<sub>H</sub> values were not different from unity. In buffer A, pK<sub>i</sub>' values for agonists remained unchanged or decreased with decreasing temperature, while antagonist pK<sub>i</sub> values increased with decreasing temperature; agonist binding was entropy-driven while antagonist binding was enthalpy and entropy-driven. In buffer A<sub>Ca</sub>, temperature had no effect on antagonist and agonist pK<sub>i</sub> values; both agonist and antagonist binding were enthalpy and entropy-driven.

**Conclusions and implications:** The binding of H<sub>3</sub>-receptor agonists and antagonists can be thermodynamically discriminated under conditions where agonist pK<sub>i</sub>' values are over-estimated (pK<sub>i</sub>' ≠ pK<sub>app</sub>). However, under conditions when agonist pK<sub>i</sub> ~ pK<sub>app</sub>, the thermodynamics underlying the binding of agonists are not different to those of antagonists.

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**Keywords:** [<sup>3</sup>H]clobenpropit; histamine H<sub>3</sub>-receptors

**Abbreviations:** PEI, polyethyleneimine; R-α-MH, R-α-methylhistamine

## Introduction

Radioligand-binding assays, second messenger assays and functional *in vitro* bioassays are used to provide receptor-affinity estimates (pK<sub>i</sub> and pK<sub>b</sub>) for novel ligands. Although these estimates are valuable in the development of high affinity, selective receptor antagonists, they provide little information about the molecular mechanisms underlying the ligand–receptor interaction. However, it is possible to obtain information about these interactions by performing thermodynamic studies in which the receptor affinity of the ligand is determined at a number of different temperatures (see Hitzeman, 1988; Raffa and Porreca, 1989).

The value of thermodynamics for investigating receptor–ligand interactions has been demonstrated in numerous studies (Weiland *et al.*, 1979; Mohler and Richards, 1981; Zahniser and Molinoff, 1983; Reith *et al.*, 1984; Kilpatrick *et al.*, 1986; Testa *et al.*, 1987; Todd and Babinski, 1987; Aronstam and Narayanan, 1988; Duarte *et al.*, 1988; Borea *et al.*, 1996a, b; Dalpiaz *et al.*, 1996; Maguire and Loew, 1996; Li *et al.*, 1998). In addition, some of these studies have suggested that measurement of thermodynamic parameters can allow the discrimination of agonist and antagonist ligands (Weiland *et al.*, 1979; Zahniser and Molinoff, 1983; Borea *et al.*, 1996a).

There has been no thermodynamic analysis of ligand binding at the histamine H<sub>3</sub>-receptor. Therefore, the primary aim of this study was to determine whether the binding of ligands, characterised as agonists and antagonists at histamine H<sub>3</sub>-receptor in a functional bioassay of the guinea-pig ileum, could be discriminated thermodynamically.

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However, in previous studies, we have shown that agonist affinity values are overestimated in H<sub>3</sub>-receptor radioligand-binding assays conducted in the absence of buffer salts by an amount which is related to the ligand's intrinsic activity (see Harper *et al.*, 2007) and, additionally, that when radioligand-binding assays are conducted in the presence of salts, pK<sub>i</sub> estimates are closer to pK<sub>app</sub> values estimated by the method of Furchgott in a guinea-pig ileum bioassay. Therefore, an additional aim of this study was to establish whether the thermodynamic parameters underlying the binding interaction of agonists paralleled those of the antagonists, when the assay was conducted under conditions where their pK<sub>i</sub> values were similar to pK<sub>app</sub> values and where the agonist  $n_H$  values were not different from unity.

The agonist ligands that were selected for this study had varying intrinsic activity ( $\alpha$ ) as defined by bioassay on the guinea-pig ileum myenteric plexus longitudinal muscle (see Figure 1 and Harper *et al.*, 2007) (immepip,  $\alpha = 1.0$ ; imetit,  $\alpha = 0.90$ ; *R*- $\alpha$ -methylhistamine (*R*- $\alpha$ -MH),  $\alpha = 1.0$ ; proxyfan,  $\alpha = 0.35$ ; chloroproxyfan  $\alpha = 0.45$ ; bromoproxyfan  $\alpha = 0.65$ ; and iodoproxyfan,  $\alpha = 0.90$ ). The antagonist ligands, as defined by bioassay (thioperamide, JB96132 and [<sup>3</sup>H]clobenpropit), all contained an imidazole moiety (Figure 1 and Harper *et al.*, 2007).

A preliminary account of some of these data was presented to the British Pharmacological Society (Harper *et al.*, 2002).

## Methods

### Preparation of guinea-pig cerebral cortex membranes

Adult male Dunkin–Hartley guinea-pigs (200–300 g) were killed by cervical dislocation and the whole brain removed and immediately placed in ice-cold 20 mM HEPES–NaOH buffer (buffer A; pH 7.4 at 21 ± 3°C). The cortex was dissected, weighed and homogenised in ice-cold buffer A (1 g 15 ml<sup>-1</sup>) using a polytron homogeniser (Kinematica AG, GmbH, Lucerne, Switzerland; PT-DA 3020/2TS; ~3 s × 3). The homogenate was centrifuged (100g, 5 min at 4°C) and the supernatants pooled and stored at 4°C. The pellets were rehomogenised in ice-cold buffer A (80 ml) and

re-centrifuged (100g, 5 min at 4°C). The supernatants were centrifuged (39 800g, 12 min at 4°C) and the final pellet was resuspended in buffer A (containing 3 mM metyrapone; at 4, 12, 21 or 30°C) to the required tissue concentration using a Teflon-in-glass homogeniser. Metyrapone was included in the assay buffer because this prevents [<sup>3</sup>H]clobenpropit binding to cytochrome P450 isoenzymes (see Harper *et al.*, 1999b).

### [<sup>3</sup>H]clobenpropit: tissue-concentration studies

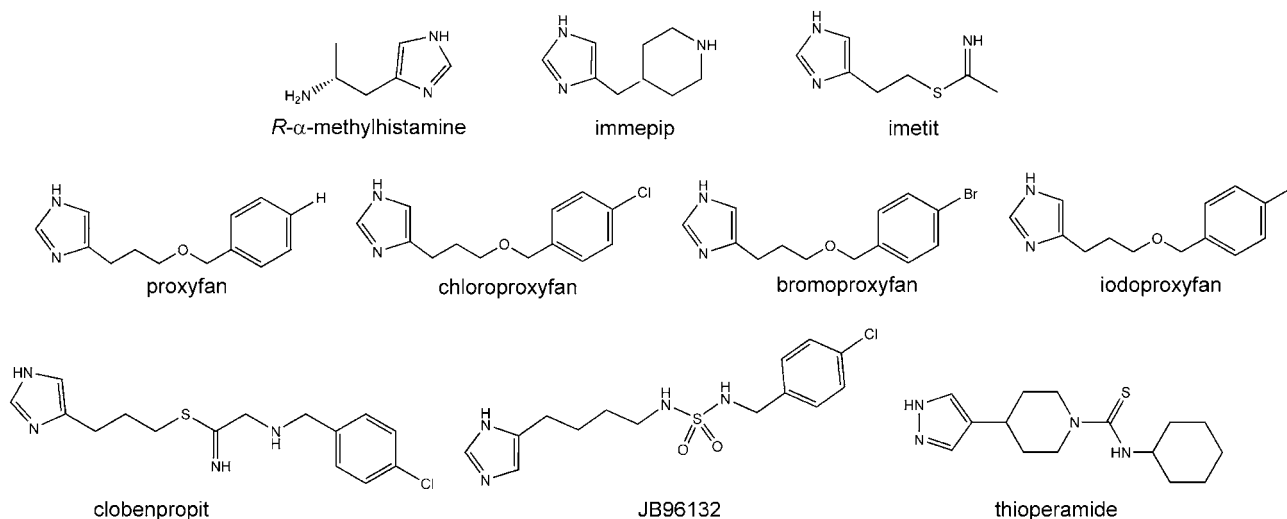
Previous studies have indicated that, when using [<sup>3</sup>H]-clobenpropit to label histamine H<sub>3</sub>-receptors in guinea-pig cerebral cortex membranes at 21°C, 1.6 mg of membranes is optimal (Harper *et al.*, 1999b). However, before investigating whether incubation temperature had any effect on the estimated affinity (pK<sub>i</sub>) of [<sup>3</sup>H]-clobenpropit, it was necessary to confirm that this was also the optimal membrane concentration for studies at 4, 12 and 30°C.

Guinea-pig cerebral cortex membranes (0.2–4 mg) were incubated (2.75 h at 21 and 30°C; 24 h at 4 and 12°C) with [<sup>3</sup>H]-clobenpropit (0.2 nM) and buffer A in a final assay volume of 0.5 ml. Total binding was defined with buffer A and non-specific binding with 1  $\mu$ M thioperamide (pK<sub>i</sub> at histamine H<sub>3</sub>-receptors in guinea-pig cortex ~9.1, Harper *et al.*, 1999a). The assay was terminated by rapid filtration through Whatman GF/B filters, pre-soaked in 0.3% polyethyleneimine (PEI), that were washed (3 × 3 ml) with ice-cold 50 mM Tris-HCl (pH 7.4 at 4°C) using a Brandell Cell Harvester. Filters were transferred into scintillation vials, 4 ml Meridian Gold-Star liquid scintillation cocktail added, and after 3 h, the bound radioactivity was determined by counting (3 min) in a Beckman liquid scintillation counter.

### [<sup>3</sup>H]clobenpropit: saturation studies

Guinea-pig cerebral cortex membranes (1.6 mg) were incubated (2.75 h at 21 and 30°C; 24 h at 4 and 12°C) in a final assay volume of 0.5 ml with buffer A and 0.004 to 3 nM [<sup>3</sup>H]clobenpropit. Total and non-specific binding were defined using buffer A and thioperamide (1  $\mu$ M).

In an additional series of experiments, we investigated the effect of temperature (4, 12, 21 and 30°C) on the binding of



**Figure 1** Structure of histamine H<sub>3</sub>-receptor agonist and antagonist ligands.

[<sup>3</sup>H]clobenpropit in the presence of 300 mM CaCl<sub>2</sub> (buffer A<sub>Ca</sub>). This was because in previous studies, we found that this buffer was more effective than 100 mM NaCl, 100 mM KCl and 70 mM CaCl<sub>2</sub> at reducing *R*-α-MH pK<sub>i</sub> values to those equivalent to pK<sub>app</sub> values estimated in a functional bioassay; see Harper *et al.*, 2007).

#### [<sup>3</sup>H]clobenpropit: kinetic studies

The observed association rate was determined at 4, 12, 21 and 30°C by incubating [<sup>3</sup>H]clobenpropit (0.2 nM) for increasing time intervals (0.25–150 min), in a final assay volume of 0.5 ml, in six tubes containing membranes (1.6 mg) and either buffer A or 1 μM thioperamide.

The dissociation rate and *t*<sub>1/2</sub> for [<sup>3</sup>H]clobenpropit was ascertained by adding 10 μl of 50 μM thioperamide to three tubes in which membranes (1.6 mg) had been incubated (4°C, 220 min; 12°C, 150 min; 21°C, 150 min; 30°C, 120 min) with [<sup>3</sup>H]clobenpropit (0.2 nM). The bound radioligand was determined at increasing incubation times (0.5–400 min).

#### Competition studies

Guinea-pig cerebral cortex membranes, resuspended in either buffer A or buffer A<sub>Ca</sub>, (1.6 mg) were incubated with [<sup>3</sup>H]clobenpropit (0.2 nM) and competing compound, in a final assay volume of 0.5 ml, for 2.75 h at 30 and 21°C and for 24 h at 12 and 4°C (at least 5 × *t*<sub>1/2</sub>; [<sup>3</sup>H]clobenpropit at each temperature). Total and non-specific binding of [<sup>3</sup>H]clobenpropit were defined using buffer A or A<sub>Ca</sub> and 1 μM thioperamide, respectively.

#### Data analysis

All data are presented as the mean ± s.e.m. unless stated otherwise.

#### Saturation data

The Hill equation was fitted to saturation data (equation (2)) using GraphPad prism software with the Hill slope (*n*<sub>H</sub>) constrained to unity and with *n*<sub>H</sub> unconstrained.

$$B = \left( \frac{B_{\max}[L]^{n_H}}{K_L^{n_H} + [L]^{n_H}} \right) \quad (1)$$

In this equation, *L* is the radioligand concentration, *B*<sub>max</sub> is the receptor density and *K*<sub>L</sub> is the equilibrium dissociation constant of the radioligand.

#### Kinetic data

Association and dissociation data were analysed using GraphPad prism software.

#### Radioligand binding: competition curve data

To obtain pIC<sub>50</sub> and *n*<sub>H</sub> parameter estimates, competition data were fitted to the Hill equation and to the Hill equation with *n*<sub>H</sub> constrained to unity, using GraphPad Prism software. Notwithstanding the finding of *n*<sub>H</sub> values that were significantly less than unity, dissociation constants (pK<sub>i</sub>) were subsequently determined from pIC<sub>50</sub> values using the Cheng and Prusoff (1973) equation to correct for the different receptor occupancy of [<sup>3</sup>H]clobenpropit in the different buffers at the different temperatures. The parameter pK<sub>i</sub>' has been assigned to dissociation constants which were derived from pIC<sub>50</sub> values, where competition curve *n*<sub>H</sub> parameter estimates were significantly less than unity. The pK<sub>L</sub> values which were used to correct pIC<sub>50</sub> values obtained in buffer A at 4, 12, 21 and 30°C are shown in Table 1 and were 10.57, 10.38, 10.40 and 10.15, respectively. The pK<sub>L</sub> values that were used to correct pIC<sub>50</sub> values obtained in buffer A<sub>Ca</sub> at 4, 12 and 21°C are shown in Table 1 and were 9.66, 9.69 and 9.77, respectively.

#### Calculation of thermodynamic parameters

The change in the standard Gibbs free energy (Δ*G*<sup>°</sup>) was calculated using the Gibbs–Helmholz thermodynamic equation (equation (2)).

$$\Delta G^\circ = -RT \ln K_A \quad (2)$$

where *R* is the ideal gas constant (8.31 J<sup>−1</sup> mol<sup>−1</sup> K), *T* is the temperature in degrees Kelvin and *K*<sub>A</sub> is the apparent association constant of the ligand at 294 K (1/*K*<sub>i</sub>).

Equation (2) is combined with the Gibbs free energy equation (3), to form the integrated van't Hoff equation (4).

$$\Delta G^\circ = \Delta H^\circ - T\Delta S^\circ \quad (3)$$

where Δ*S*<sup>°</sup> is the entropy change of binding.

$$\ln K_A = -\Delta H^\circ / RT + \Delta S^\circ / R \quad (4)$$

A plot, therefore, of ln *K*<sub>A</sub> versus 1/*T* allows the estimation of the enthalpy of binding (Δ*H*<sup>°</sup>) because the slope is −Δ*H*<sup>°</sup>/*R*. In addition, the entropy change (Δ*S*<sup>°</sup>) can be estimated as the *y*-intercept (−Δ*S*<sup>°</sup>/*R*) (see Borea *et al.*, 1998). In these studies, Δ*G*<sup>°</sup>, Δ*H*<sup>°</sup> and Δ*S*<sup>°</sup> have been designated by their primed counterparts

**Table 1** Effect of temperature and assay buffer on the estimated pK<sub>L</sub>, *n*<sub>H</sub> and *B*<sub>max</sub> of [<sup>3</sup>H]clobenpropit in guinea-pig cerebral cortex membranes

Temperature (°C)	Buffer A			Buffer A <sub>Ca</sub>		
	pK <sub>L</sub>	<i>B</i> <sub>max</sub>	<i>n</i> <sub>H</sub>	pK <sub>L</sub>	<i>B</i> <sub>max</sub>	<i>n</i> <sub>H</sub>
30	10.15 ± 0.04	4.00 ± 1.17	0.93 ± 0.02	ND	ND	ND
21	10.40 ± 0.08	3.80 ± 0.97	1.08 ± 0.09	9.77 ± 0.20	3.28 ± 0.65	1.20 ± 0.05
12	10.38 ± 0.07	4.06 ± 0.55	0.90 ± 0.03	9.69 ± 0.18	4.02 ± 0.83	1.05 ± 0.09
4	10.57 ± 0.03	4.00 ± 0.48	1.11 ± 0.06	9.66 ± 0.27	4.04 ± 0.68	1.13 ± 0.12

Abbreviations: ND, not determined.

pK<sub>L</sub>, *B*<sub>max</sub> (fmol mg<sup>−1</sup>) and *n*<sub>H</sub> values were obtained by fitting saturation data to the Hill equation. Data are the mean ± s.e.m. of three experiments.

( $\Delta G^\circ$ ,  $\Delta H^\circ$  and  $\Delta S^\circ$ ) because the studies were conducted at pH 7.4 (although it is common to see the primes omitted). This is because the terms  $\Delta G^\circ$ ,  $\Delta H^\circ$  and  $\Delta S^\circ$ , only apply to measurements made under standard state conditions of 1 atmosphere and unit activity (sometimes stated as 1 M concentration) and at a 1 M hydrogen ion concentration (pH 0).

#### Statistical analysis

Differences in  $pK_L$  and  $B_{\max}$  values of [<sup>3</sup>H]clobenpropit were determined using analysis of variance (ANOVA) with Bonferro-ni *post hoc* test or paired *t*-test. An *F*-test was used to establish whether saturation data was best fitted by the Hill equation or by the Hill equation with  $n_H$  constrained to unity. The significance of differences in  $pK_i$  values obtained at different temperatures, in replicate experiments, was determined by paired *t*-test.  $r^2$  values obtained from linear regression were used to determine whether there was a significant linear relationship between temperature ( $1/T$ ) and  $\ln K_A$  values. An *F*-test was used to establish whether the slope was significantly different from zero. *P*-values of less than 0.05 were considered significant.

#### Materials

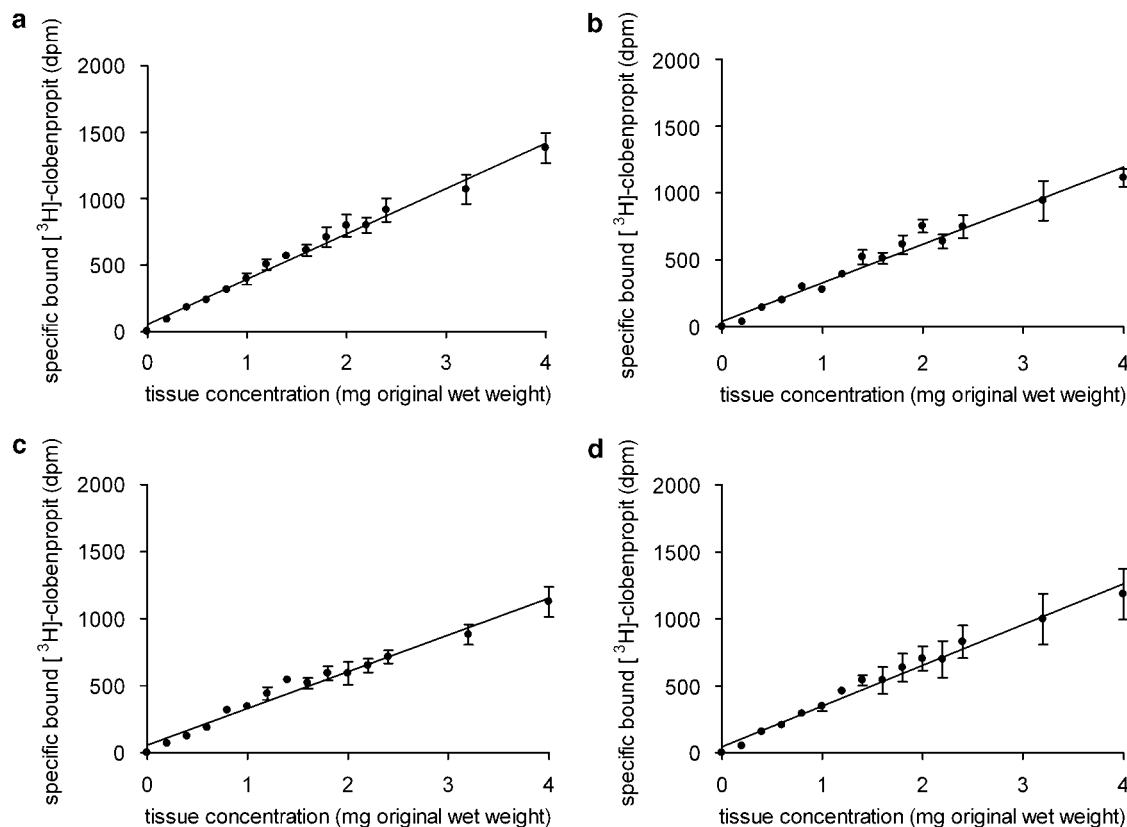
[<sup>3</sup>H]clobenpropit (VUF9153) was prepared by Amersham International plc, (Little Chalfont, Buckinghamshire, UK) to a specific activity of 45 Ci mmol<sup>-1</sup>.

Proxyfan, chloroproxyfan, bromoproxyfan, iodoproxyfan, immepip and JB96132 were synthesised by James Black Foundation Chemists. 2-Methyl-1,2-di-3-pyridyl-1-propa-none (metyrapone), HEPES (*N*-[2-hydroxyethyl]piperazine-*N'*-[2-ethanesulphonic acid]) and Trizma base were obtained from Sigma Chemical Co., (Poole, Dorset, UK). *R*- $\alpha$ -MH, thioperamide and imetit were obtained from Research Biochemicals Inc., (Poole, Dorset, UK). All other materials were obtained from Fisher Scientific, Loughborough (Leices-tershire, UK).

## Results

#### Effect of incubation temperature on the optimal tissue concentration

At 4, 12, 21 and 30°C (277, 285, 294 and 303 K), there was a linear relationship between the specific binding of 0.2 nM [<sup>3</sup>H]clobenpropit and membrane concentration up to 4 mg (see Figure 2). In buffer A, at 4, 12, 21 and 30°C, and a membrane concentration of 1.6 mg,  $14.0 \pm 1.7$ ;  $13.1 \pm 1.3$ ;  $13.3 \pm 1.0$  and  $11.9 \pm 0.9\%$  of the added [<sup>3</sup>H]clobenpropit was bound, respectively ( $n=3$ ). There was no significant difference between the specific binding of [<sup>3</sup>H]clobenpropit as a percent of total binding (percentage specific binding) at each temperature in buffer A (4°C =  $41.5 \pm 3.4$ ; 12°C =  $38.0 \pm 4.9$ ; 21°C =  $38.8 \pm 5.0$ ; 30°C =  $42.9 \pm 8.6\%$ ;  $n=3$ , ANOVA).



**Figure 2** Linearity of the relationship between specific binding of [<sup>3</sup>H]clobenpropit (0.2 nM) and added guinea-pig cerebral cortex membrane concentration at (a) 4°C, (b) 12°C, (c) 21°C and (d) 30°C. Increasing concentrations of guinea-pig cerebral cortex membranes (0.2–4 mg) were incubated in triplicate with [<sup>3</sup>H]clobenpropit (0.2 nM) and buffer A in a final assay volume of 0.5 ml. The incubation was terminated after 2.75 h at 21 and 30°C and after 24 h at 4 and 12°C. Total binding was defined with buffer A and non-specific binding with 1  $\mu$ M thioperamide. Data represent the mean  $\pm$  s.e.m. of three experiments. The line shown superimposed on the data was obtained by linear regression.

In buffer A<sub>Ca</sub>, at 4, 12 and 21°C, and a membrane concentration of 1.6 mg,  $5.2 \pm 0.8$ ;  $4.5 \pm 0.8$  and  $5.2 \pm 0.8\%$  of the added 0.2 nM [<sup>3</sup>H]clobenpropit was bound, respectively ( $n=8$ ). The percentage specific binding of [<sup>3</sup>H]clobenpropit at 21°C was significantly lower than that obtained at 4 and 12°C ( $4^\circ\text{C}=77.9 \pm 2.6$ ;  $12^\circ\text{C}=76.3 \pm 2.5$  and  $21^\circ\text{C}=66.3 \pm 2.8\%$ ;  $n=4$ ,  $P<0.05$ , ANOVA and Bonferroni *post hoc* test) and, at all temperatures, the non-specific binding was lower than that obtained in buffer A.

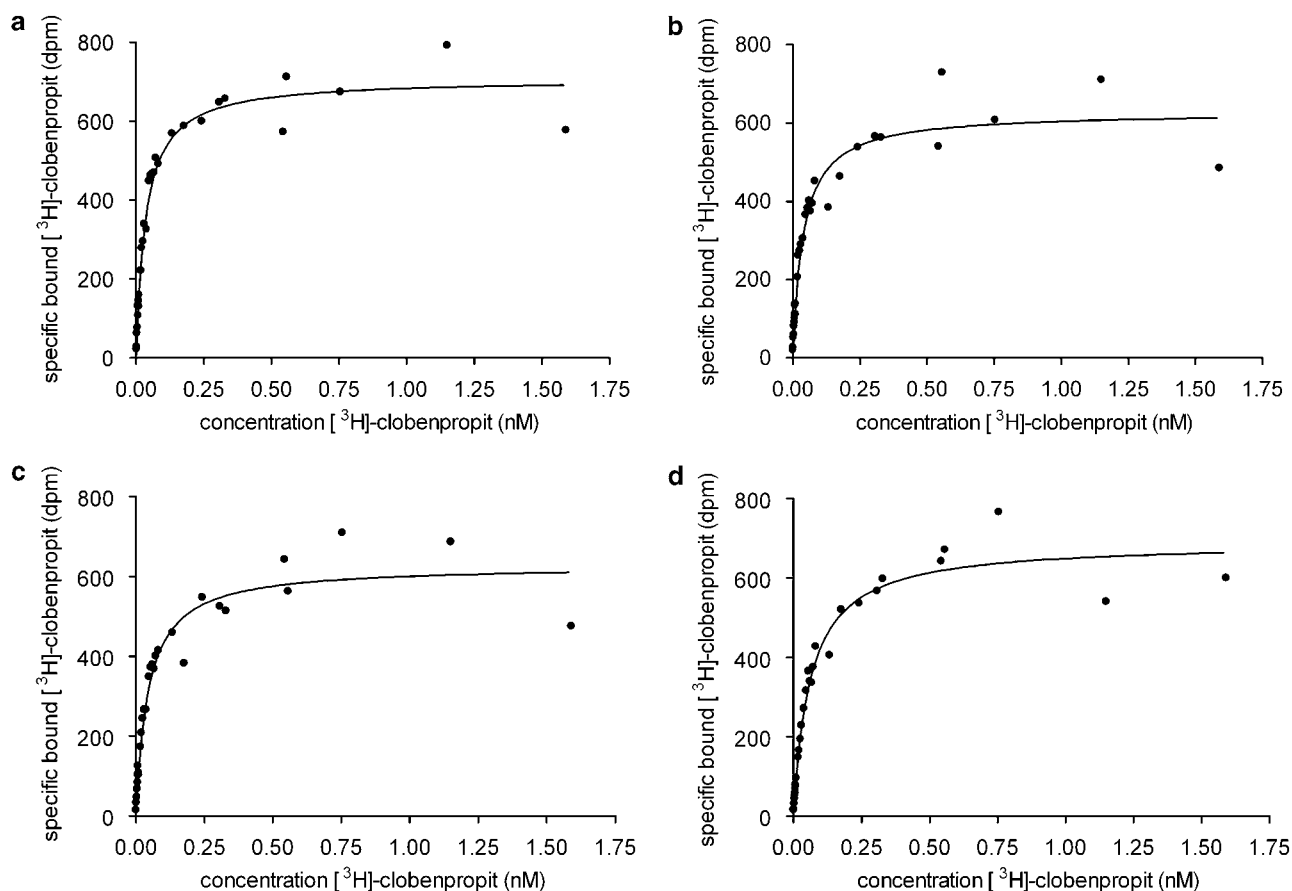
A membrane concentration of 1.6 mg was used for all subsequent experiments in both buffers.

#### Effect of incubation temperature on [<sup>3</sup>H]clobenpropit saturation isotherms

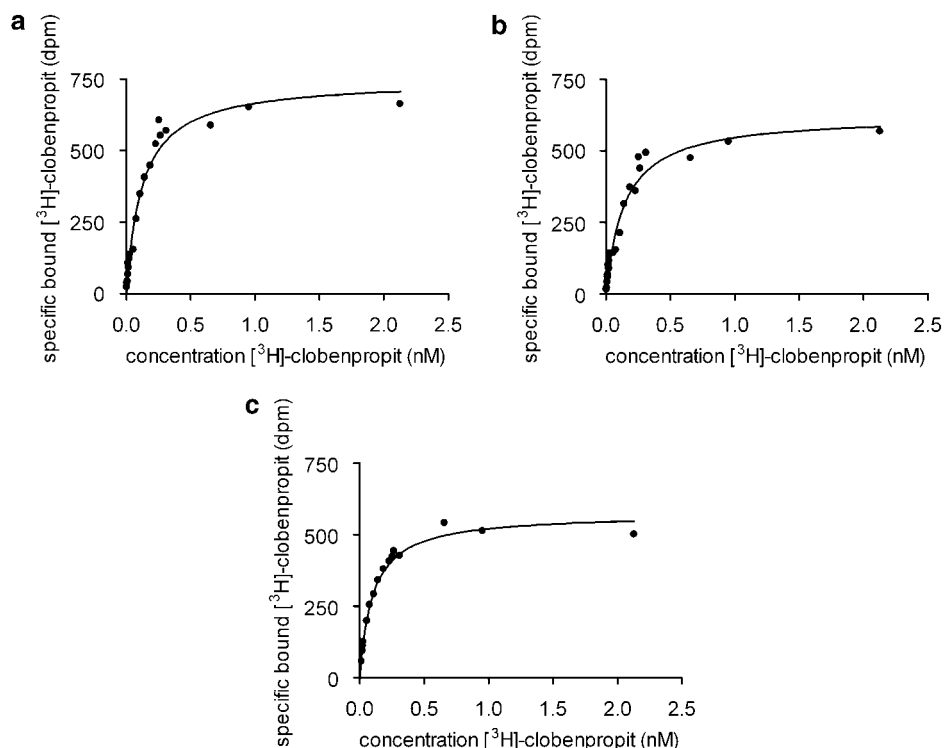
In buffer A, the binding of [<sup>3</sup>H]clobenpropit was saturable at all temperatures (Figure 3). The mean Hill slope parameter estimates ( $n_H$ ), at the four temperatures, were not significantly different from unity (Table 1, *t*-test) and in each experiment, for each temperature, there was no significant difference between the fit to the Hill equation and the fit to the Hill equation with  $n_H$  constrained to unity ( $P>0.05$ , *F*-test). The incubation temperature had no significant effect on the estimated histamine H<sub>3</sub>-receptor density ( $B_{\text{max}}$ , Table 1, ANOVA).

In buffer A<sub>Ca</sub>, at 30°C and over the concentration range of [<sup>3</sup>H]clobenpropit used (0.004–3 nM), the specific binding was lower and more variable than that obtained at 4, 12 and 21°C and, as a result, it was not possible to obtain accurate  $B_{\text{max}}$  or  $pK_L$  estimates. At 21, 12 and 4°C, the specific binding of [<sup>3</sup>H]clobenpropit was saturable (Figure 4), mean Hill slope parameter estimates ( $n_H$ ), were not significantly different from unity (Table 1, *t*-test) and in each experiment for each temperature, there was no significant difference between the fit to the Hill equation and the fit to the Hill equation with  $n_H$  constrained to unity ( $P>0.05$ , *F*-test). The incubation temperature did not significantly change the estimated H<sub>3</sub>-receptor  $B_{\text{max}}$  (Table 1, ANOVA).

In buffer A, the  $pK_L$  of [<sup>3</sup>H]clobenpropit increased significantly with decreasing incubation temperature (Table 1, ANOVA and Bonferroni *post hoc* test,  $P<0.05$ ;) and the  $pK_L$  was significantly higher at 4°C than at 30°C (paired *t*-test,  $P<0.05$ ). A van't Hoff plot of  $\ln K_A$  versus  $1/T$  was linear with a positive slope which was significantly different from zero (*F*-test,  $P<0.05$ ; Figure 5). In contrast, in buffer A<sub>Ca</sub>, there was no relationship between assay incubation temperature and  $pK_L$ , and there was no significant difference in the  $pK_L$  of [<sup>3</sup>H]clobenpropit at the three temperatures (Table 1, paired *t*-test). The slope of the van't Hoff plot for



**Figure 3** Saturation analysis of the binding of [<sup>3</sup>H]clobenpropit to guinea-pig cerebral cortex H<sub>3</sub>-receptors at (a) 4°C, (b) 12°C, (c) 21°C and (d) 30°C, in buffer A. Tissue (1.6 mg) was incubated in triplicate with increasing concentrations of [<sup>3</sup>H]clobenpropit (0.004–3 nM) in a final assay volume of 0.5 ml. Total and non-specific binding were defined with buffer A or 1  $\mu\text{M}$  thioperamide, respectively. The incubation was terminated after 2.75 h at 21 and 30°C and after 24 h at 4 and 12°C. Data are representative of three experiments. The line shown superimposed through the data are the fit to the Hill equation.



**Figure 4** Saturation analysis of the binding of [<sup>3</sup>H]clobenpropit to guinea-pig cerebral cortex H<sub>3</sub>-receptors at (a) 4°C, (b) 12°C and (c) 21°C, in buffer A<sub>Ca</sub>. Tissue (1.6 mg) was incubated in triplicate with increasing concentrations of [<sup>3</sup>H]clobenpropit (0.004–3 nM). Total and non-specific binding were defined with buffer A<sub>Ca</sub> or 1  $\mu$ M thioperamide, respectively. The incubation was terminated after 2.75 h at 21°C and after 24 h at 4 and 12°C. Data is representative of three experiments.

[<sup>3</sup>H]clobenpropit, in buffer A<sub>Ca</sub>, was not different from zero (*F*-test, Figure 5).

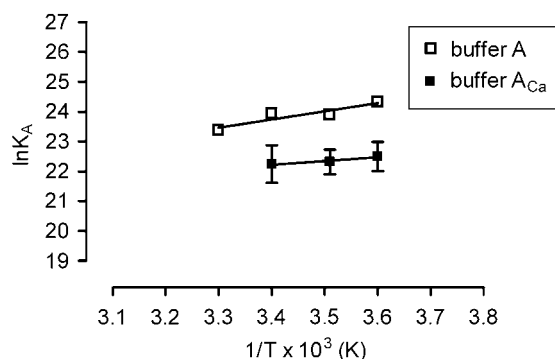
*Effect of incubation temperature on the association and dissociation rates of [<sup>3</sup>H]clobenpropit*

In buffer A, the specific binding of 0.2 nM [<sup>3</sup>H]clobenpropit reached equilibrium after approximately 80, 30, 25 and 3 min incubations at 4, 12, 21 and 30°C, respectively (*n* = 3, Figure 6). The [<sup>3</sup>H]clobenpropit association data obtained at all four temperatures could be fitted by a pseudo-first-order rate equation. The association rate constants (*k*<sub>+1</sub>) for the binding of [<sup>3</sup>H]clobenpropit to histamine H<sub>3</sub>-receptors on guinea-pig cortex membranes at 4, 12, 21 and 30°C are given in Table 2.

The dissociation data for [<sup>3</sup>H]clobenpropit could also be fitted by a first-order rate equation. The dissociation rate constants (*k*<sub>-1</sub>) at 4, 12, 21 and 30°C are shown in Table 2, along with the calculated *pK<sub>L</sub>* values; the latter increased with decreasing temperature (Table 2). These *pK<sub>L</sub>* values were not significantly different from those obtained using saturation analysis at 4, 12, 21 and 30°C (compare *pK<sub>L</sub>* values in Tables 1 and 2, *t*-test).

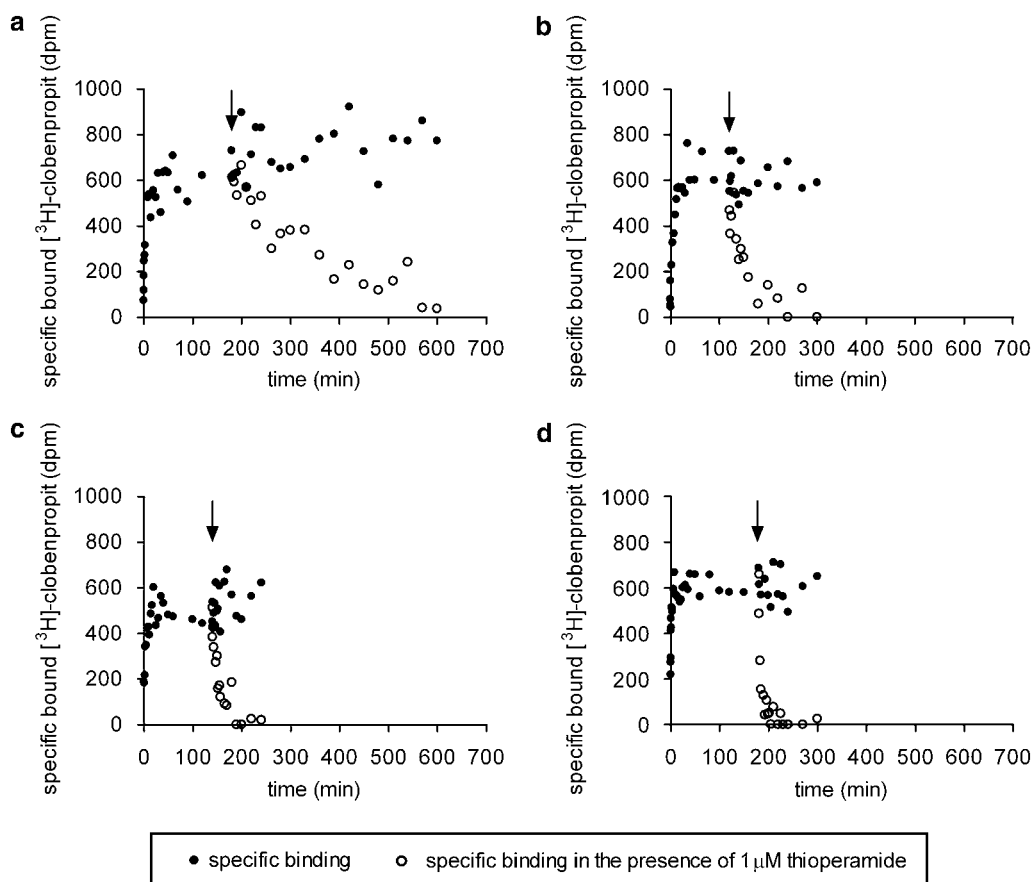
*Effect of incubation temperature on ligand *pK<sub>i</sub>* values in buffer A and buffer A<sub>Ca</sub>*

In both buffer A and buffer A<sub>Ca</sub>, each histamine H<sub>3</sub>-receptor ligand produced a concentration-dependent inhibition of



**Figure 5** Van't Hoff plot showing the effect of temperature on the equilibrium association constants (*K<sub>A</sub>*) of [<sup>3</sup>H]clobenpropit in 20 mM HEPES buffer (buffer A) and in buffer A with 300 mM CaCl<sub>2</sub> (buffer A<sub>Ca</sub>). Data are the mean  $\pm$  s.e.m. of three replicate experiments. The lines shown superimposed through the data were obtained by linear regression.

the specific binding of [<sup>3</sup>H]clobenpropit to H<sub>3</sub>-receptors in guinea-pig cerebral cortex membranes at all temperatures, as illustrated in Figure 7 for *R*- $\alpha$ -MH and thioperamide. In buffer A, the mean mid-point slope parameter estimates (*n<sub>H</sub>*) for all the agonists, except proxyfan and imetit (see Table 3), at most incubation temperatures were significantly less than unity (*t*-test, *P* < 0.05). The mean mid-point slope parameter estimate for thioperamide was significantly less than unity at 21 and 4°C (Table 3, *t*-test, *P* < 0.05).



**Figure 6** Representative association–dissociation analysis of [<sup>3</sup>H]clobenpropit binding to H<sub>3</sub>-receptors in guinea-pig cerebral cortex membranes at (a) 4°C, (b) 12°C, (c) 21°C and (d) 30°C. The association rate was determined under pseudo-first-order conditions because only ~10% of the added [<sup>3</sup>H]clobenpropit was bound. [<sup>3</sup>H]clobenpropit (0.2 nM) was incubated, in a final assay volume of 0.5 ml for increasing times with membranes (1.6 mg at 4, 12, 21 or 30°C). Total and non-specific binding were defined with buffer A and 1 μM thioperamide, respectively. The dissociation rate for [<sup>3</sup>H]clobenpropit from H<sub>3</sub>-receptors was determined by incubating [<sup>3</sup>H]clobenpropit (0.2 nM) with membranes and buffer A and then adding 10 μl of 50 μM thioperamide (arrow). The bound radioligand was determined at increasing incubation times.

**Table 2** Effect of temperature on the association rate, dissociation rate and pK<sub>L</sub> of [<sup>3</sup>H]clobenpropit at histamine H<sub>3</sub>-receptors in guinea-pig cerebral cortex membranes

Temperature (°C)	Temperature (K)	k <sub>+1</sub> (M <sup>-1</sup> min <sup>-1</sup> )	k <sub>-1</sub> (min <sup>-1</sup> )	pK <sub>L</sub>
30	303	77.26 ± 5.29 × 10 <sup>8</sup>	3.32 ± 0.50 × 10 <sup>-1</sup>	10.37 ± 0.10
21	294	7.49 ± 1.32 × 10 <sup>8</sup>	0.44 ± 0.06 × 10 <sup>-1</sup>	10.27 ± 0.27
12	285	10.60 ± 0.84 × 10 <sup>8</sup>	0.33 ± 0.05 × 10 <sup>-1</sup>	10.60 ± 0.12
4	277	6.37 ± 3.3 × 10 <sup>8</sup>	0.047 ± 0.001 × 10 <sup>-1</sup>	11.01 ± 0.20

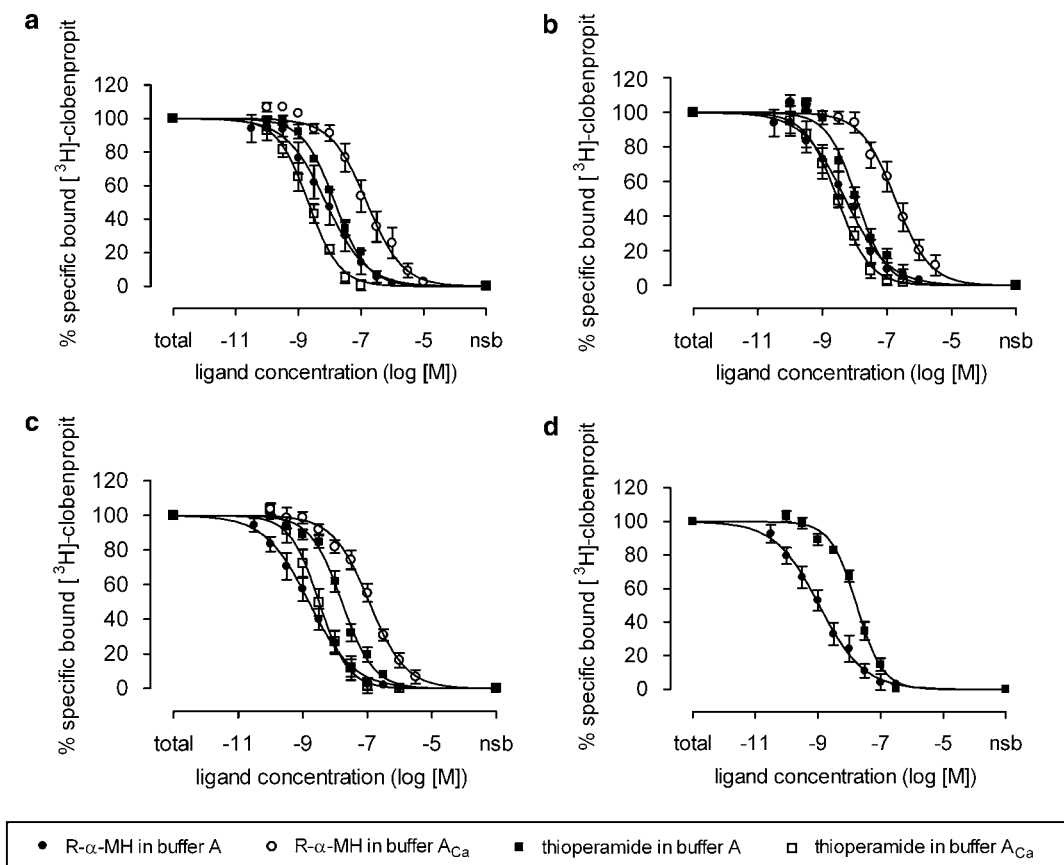
Data are the mean ± s.e.m. of three experiments.

In buffer A<sub>Ca</sub>, the mean *n<sub>H</sub>* parameter estimates for the agonists were not significantly different from unity with the exception of iodoproxyfan and *R*-α-MH at 12°C and imipip at 4 and 12°C. The mean *n<sub>H</sub>* value for thioperamide was significantly less than unity at 4°C (*t*-test, *P* < 0.05).

Notwithstanding the finding of *n<sub>H</sub>* values that were significantly less than unity, dissociation constants were subsequently determined from pIC<sub>50</sub> values using the Cheng and Prusoff (1973) equation to correct for the different receptor occupancy of [<sup>3</sup>H]clobenpropit in the different buffers at the different temperatures. The parameter pK<sub>I</sub>'

has been assigned to dissociation constants which were derived from pIC<sub>50</sub> values, where mean *n<sub>H</sub>* parameter estimates were significantly less than unity. The pK<sub>L</sub> values that were used to correct pIC<sub>50</sub> values obtained in buffer A and buffer A<sub>Ca</sub> are presented in Table 1.

The effect of temperature on pK<sub>I</sub> or pK<sub>I</sub>' values of all ligands in buffer A and buffer A<sub>Ca</sub> are listed in Table 3. In buffer A, the pK<sub>I</sub> values of the two antagonist ligands, thioperamide and JB96132 were significantly higher at 4°C than at 30°C (*P* < 0.05, paired *t*-test). The pK<sub>I</sub>' values for the H<sub>3</sub>-receptor agonists, proxyfan, chloroproxyfan, iodoprox-



**Figure 7** Competition curves for an H<sub>3</sub>-receptor agonist, *R*- $\alpha$ -MH, and antagonist, thioperamide, at sites labelled with 0.2 nM [<sup>3</sup>H]clobenpropit in guinea-pig cerebral cortex. (a–c) Mean % specific binding of *R*- $\alpha$ -MH and thioperamide at 4°C, 12 and 21°C in buffer A and buffer A<sub>Ca</sub> (d) mean % specific binding of *R*- $\alpha$ -MH and thioperamide at 30°C in buffer A. Guinea-pig cerebral cortex membranes (1.6 mg) were incubated in a final volume of 0.5 ml with HEPES–NaOH buffer, [<sup>3</sup>H]clobenpropit (0.2 nM) and increasing concentrations of ligands for 2.75 h at 30 and 21°C and for 24 h at 12 and 4°C. Total and non-specific binding of [<sup>3</sup>H]clobenpropit were defined using appropriate buffer and 1  $\mu$ M thioperamide, respectively. Data are the mean  $\pm$  s.e.m. of between five and six experiments (see Table 3). The lines shown superimposed on the data were obtained using the fit to the Hill equation.

yfan and *R*- $\alpha$ -MH were significantly lower at 4°C than at 30°C ( $P < 0.05$ , paired *t*-test) and there was no significant difference between  $pK_i'$  values obtained for imipip, imetit and bromopropyfan at these temperatures (paired *t*-test). In buffer A<sub>Ca</sub>, there was no significant difference between  $pK_i$  values at 4 and 21°C for all ligands (Table 3, paired *t*-test).

#### Thermodynamic parameters of ligand binding

Van't Hoff plots of  $\ln K_A$  versus  $1/T$  were constructed for all ligands using the affinity values ( $1/K_i$ ) obtained in buffer A (4, 12, 21 and 30°C) and buffer A<sub>Ca</sub> (4, 12, 21°C) (see, for e.g., Figures 8 and 9). The van't Hoff plots, constructed for thioperamide and JB96132, using  $pK_i$  values obtained in buffer A, had positive slopes which were significantly different from zero (Figure 8, *F*-test,  $P < 0.05$ ). The van't Hoff plots, constructed for proxyfan, chloropropyfan and iodo-proxyfan using  $pK_i$  values obtained in buffer A, had negative slopes which were significantly different from zero (*F*-test,  $P < 0.05$ ). The van't Hoff plots, constructed for all ligands using  $pK_i$  values obtained in buffer A<sub>Ca</sub>, had slopes which were not significantly different from zero (*F*-test,  $P < 0.05$ ).

Mean values of enthalpy ( $\Delta H^\circ$ ), entropy ( $\Delta S^\circ$ ),  $-T\Delta S^\circ$  and the Gibbs free energy ( $\Delta G^\circ$ ) of ligands in buffer A and A<sub>Ca</sub>, were obtained from van't Hoff plots and are presented in Tables 4 and 5.  $\Delta G^\circ$  was also calculated at 21°C (294 K) using the Gibbs–Helmholz equation.

In buffer A, the mean  $\Delta H^\circ$  values for JB96132, thioperamide and [<sup>3</sup>H]clobenpropit were negative (Table 4) and mean  $\Delta S^\circ$  values were positive (Table 4). The mean  $\Delta H^\circ$  and  $\Delta S^\circ$  values for the ligands classified as agonists in the guinea-pig ileum longitudinal muscle myenteric plexus (see Harper *et al.*, 2007) were also positive (Table 4).

In buffer A<sub>Ca</sub>, the mean  $\Delta H^\circ$  values for the antagonists were all negative, over a fourfold range, while the mean  $\Delta S^\circ$  values were positive (Table 5). The mean  $\Delta H^\circ$  values for the agonist ligands, with the exception of imipip and imetit, were also negative (Table 5) and the mean  $\Delta S^\circ$  values for the agonists in buffer A<sub>Ca</sub> were positive (Table 5).

#### Extrathermodynamic plots

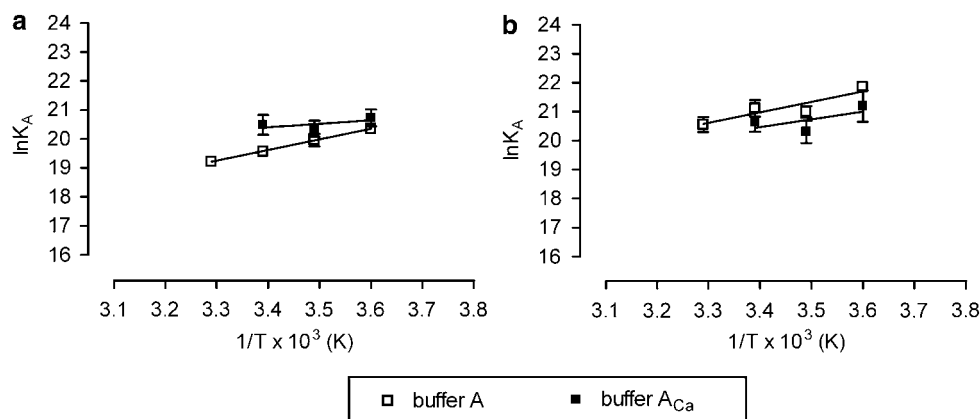
A plot of the  $\Delta H^\circ$  and  $-T\Delta S^\circ$  for all ligands, obtained in buffer A and shown in Table 3, indicated that the binding of the antagonists was enthalpy- and entropy-driven (Figure 10a).

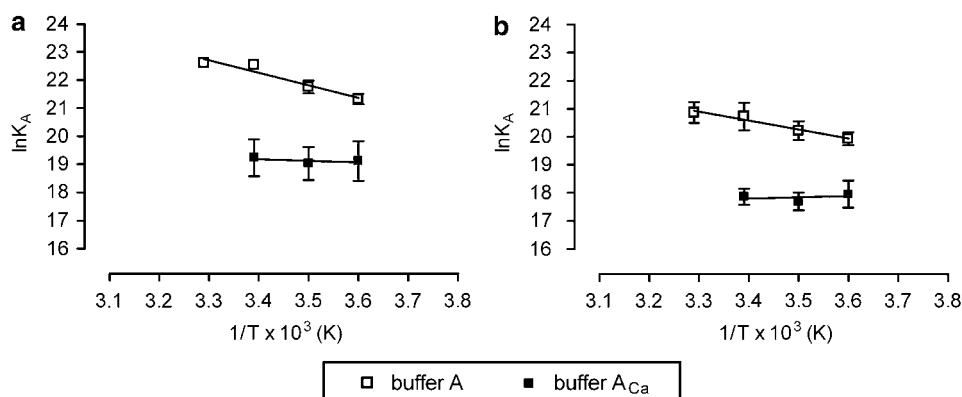


**Table 3** pK<sub>i</sub> and n<sub>H</sub> values for histamine H<sub>3</sub>-receptor agonists and antagonists at 4, 12, 21 and 30°C (277, 285, 294 and 300 K)

Ligand	Temperature (K)	Buffer A			Buffer A <sub>Ca</sub>		
		pK <sub>i</sub> or pK' <sub>i</sub>	n <sub>H</sub>	n	pK <sub>i</sub> or pK' <sub>i</sub>	n <sub>H</sub>	n
Histamine H <sub>3</sub> -receptor agonists							
proxifyan	303	8.51 ± 0.13	0.97 ± 0.21	3	ND	ND	
	294	8.61 ± 0.17	0.73 ± 0.06		ND	ND	
	285	8.05 ± 0.08	0.89 ± 0.06		ND	ND	
chloroproxyfan	277	7.89 ± 0.07	1.07 ± 0.12	5	ND	ND	4
	303	9.04 ± 0.15	0.81 ± 0.04 <sup>a</sup>		ND	ND	
	294	9.00 ± 0.22	0.77 ± 0.04 <sup>a</sup>		7.76 ± 0.12	1.04 ± 0.05	
	285	8.78 ± 0.14	0.71 ± 0.07 <sup>a</sup>		7.68 ± 0.14	0.88 ± 0.04	
	277	8.66 ± 0.10	0.90 ± 0.12		7.82 ± 0.20	0.94 ± 0.07	
bromoproxyfan	303	9.32 ± 0.05	0.77 ± 0.07 <sup>a</sup>	3	ND	ND	4
	294	9.28 ± 0.02	0.83 ± 0.09		8.13 ± 0.26	0.99 ± 0.04	
	285	9.04 ± 0.14	0.81 ± 0.04 <sup>a</sup>		8.04 ± 0.30	0.91 ± 0.06	
	277	9.13 ± 0.15	0.88 ± 0.16		8.16 ± 0.29	0.95 ± 0.08	
	303	9.82 ± 0.07	0.65 ± 0.10 <sup>a</sup>		ND	ND	
iodoproxyfan	294	9.79 ± 0.05	0.67 ± 0.03 <sup>a</sup>	4	8.10 ± 0.36	0.94 ± 0.08	4
	285	9.45 ± 0.10	0.62 ± 0.08 <sup>a</sup>		8.26 ± 0.26	0.88 ± 0.02	
	277	9.27 ± 0.08	0.94 ± 0.16		8.31 ± 0.31	0.91 ± 0.03	
	303	9.17 ± 0.19	0.76 ± 0.09		ND	ND	
	294	9.33 ± 0.12	0.80 ± 0.12		7.93 ± 0.17	0.86 ± 0.05	
imetit	285	9.03 ± 0.07	0.85 ± 0.08	4	7.87 ± 0.12	0.89 ± 0.05	6
	277	9.16 ± 0.12	0.87 ± 0.08		7.89 ± 0.14	0.94 ± 0.07	
	303	9.13 ± 0.22	0.63 ± 0.06 <sup>a</sup>		ND	ND	
	294	9.15 ± 0.18	0.63 ± 0.02 <sup>a</sup>		7.30 ± 0.14	0.83 ± 0.07	
	285	8.89 ± 0.21	0.66 ± 0.05 <sup>a</sup>		7.09 ± 0.20	0.81 ± 0.05 <sup>a</sup>	
R-α-MH	277	8.87 ± 0.22	0.62 ± 0.03 <sup>a</sup>	5	7.21 ± 0.22	0.86 ± 0.04	5
	303	9.47 ± 0.03	0.71 ± 0.06 <sup>a</sup>		ND	ND	
	294	9.53 ± 0.04	0.88 ± 0.05		8.07 ± 0.11	0.86 ± 0.05	
	285	9.39 ± 0.13	0.82 ± 0.05 <sup>a</sup>		7.91 ± 0.12	0.91 ± 0.03 <sup>a</sup>	
	277	9.34 ± 0.14	0.73 ± 0.03 <sup>a</sup>		7.99 ± 0.19	0.90 ± 0.02 <sup>a</sup>	
Histamine H <sub>3</sub> -receptor antagonists							
JB96132	303	8.95 ± 0.07	1.08 ± 0.14	3	ND	ND	4
	294	9.17 ± 0.13	0.99 ± 0.06		8.97 ± 0.14	1.06 ± 0.12	
	285	9.11 ± 0.09	0.85 ± 0.05		8.82 ± 0.17	0.95 ± 0.04	
	277	9.49 ± 0.05	0.97 ± 0.09		9.20 ± 0.24	0.94 ± 0.03	
thiopiperamide	303	8.35 ± 0.06	0.90 ± 0.08	6	ND	ND	5
	294	8.50 ± 0.05	0.82 ± 0.04 <sup>a</sup>		9.00 ± 0.15	0.96 ± 0.11	
	285	8.66 ± 0.09	0.96 ± 0.06		8.84 ± 0.12	0.88 ± 0.05	
	277	8.75 ± 0.02	0.81 ± 0.04 <sup>a</sup>		9.00 ± 0.13	0.86 ± 0.03 <sup>a</sup>	

Abbreviation: ND, not determined.

Data are the mean ± s.e.m. pK'<sub>i</sub> is the ligand affinity when n<sub>H</sub> is significantly less than unity.<sup>a</sup>n<sub>H</sub> significantly different from unity, P < 0.05 t-test.**Figure 8** Van't Hoff plots showing the effect of temperature on the equilibrium association constants ( $K_A$ ) of (a) thiopiperamide and (b) JB96132 in buffer A and buffer A<sub>Ca</sub>. Data are the mean ± s.e.m. of between three and six replicate experiments. The lines shown superimposed through the data were obtained by linear regression.



**Figure 9** Van't Hoff plots showing the effect of temperature on the equilibrium association constants ( $K_A$ ) of (a) iodoproxyfan and (b) chloroproxyfan in buffers A and  $A_{Ca}$ . Data are the mean  $\pm$  s.e.m. of between four and five replicate experiments. The lines shown superimposed through the data were obtained by linear regression.

**Table 4** Thermodynamic parameters of binding to histamine H<sub>3</sub>-receptors in guinea-pig cerebral cortex membranes in buffer A

Ligand	Intrinsic activity ( $\alpha$ )	$\Delta G^{or}$ calculated ( $\text{kJ mol}^{-1}$ ) <sup>a</sup>	$\Delta G^{or}$ ( $\text{kJ mol}^{-1}$ )	$\Delta H^{or}$ ( $\text{kJ mol}^{-1}$ )	$-T\Delta S^{or}$ ( $\text{kJ mol}^{-1}$ )	$\Delta S^{or}$ ( $\text{J mol}^{-1} \text{K}^{-1}$ )	n
<i>Histamine H<sub>3</sub>-receptor agonists</i>							
imnepip	1.00 <sup>b</sup>	$-53.8 \pm 0.3$	$-53.4 \pm 0.2$	$9.1 \pm 10.0$	$-62.5 \pm 9.9$	$211.7 \pm 33.3$	5
imetit	0.90 <sup>b,c</sup>	$-52.6 \pm 0.7$	$-51.9 \pm 0.8$	$6.4 \pm 6.2$	$-58.3 \pm 6.9$	$197.8 \pm 23.3$	4
R- $\alpha$ -MH	1.00 <sup>c</sup>	$-51.6 \pm 1.0$	$-51.2 \pm 1.1$	$19.1 \pm 3.9$	$-70.3 \pm 4.0$	$218.3 \pm 14.2$	5
proxifyan	0.35 <sup>c</sup>	$-48.6 \pm 1.0$	$-47.4 \pm 0.7$	$44.5 \pm 9.2$	$-91.8 \pm 9.8$	$311.3 \pm 33.1$	3
chloroproxyfan	0.45 <sup>c</sup>	$-46.9 \pm 2.9$	$-50.5 \pm 0.9$	$26.6 \pm 4.9$	$-77.1 \pm 5.8$	$261.4 \pm 19.7$	5
bromoproxyfan	0.69 <sup>c</sup>	$-51.6 \pm 0.9$	$-52.2 \pm 0.2$	$14.2 \pm 13.4$	$-66.3 \pm 13.3$	$224.9 \pm 45.0$	3
iodoproxyfan	0.90 <sup>c</sup>	$-55.3 \pm 0.3$	$-54.7 \pm 0.3$	$37.3 \pm 7.3$	$-92.0 \pm 7.3$	$311.0 \pm 24.4$	4
<i>Histamine H<sub>3</sub>-receptor antagonists</i>							
JB96132	0 <sup>c</sup>	$-51.8 \pm 0.7$	$-51.0 \pm 0.5$	$-34.1 \pm 6.5$	$-16.9 \pm 6.8$	$57.3 \pm 23.2$	3
thioperamide	0 <sup>c</sup>	$-47.8 \pm 0.3$	$-48.0 \pm 0.3$	$-30.5 \pm 6.1$	$-17.5 \pm 6.1$	$59.3 \pm 20.6$	6
[ <sup>3</sup> H]clobenpropit	0 <sup>c</sup>	$-58.5 \pm 0.4$	$-58.0 \pm 0.3$	$-22.9 \pm 1.2$	$-35.1 \pm 1.3$	$119.5 \pm 4.4$	3

Data are the mean  $\pm$  s.e.m.

<sup>a</sup> $\Delta G^{or}$  at 21°C, calculated using equation 2.

<sup>b</sup> $\alpha$  values from Alves-Rodrigues *et al.*, 2001

<sup>c</sup>Intrinsic activity in guinea-pig ileum bioassay, see Harper *et al.* (2007).

**Table 5** Thermodynamic parameters of binding to histamine H<sub>3</sub>-receptors in guinea-pig cerebral cortex membranes in buffer  $A_{Ca}$

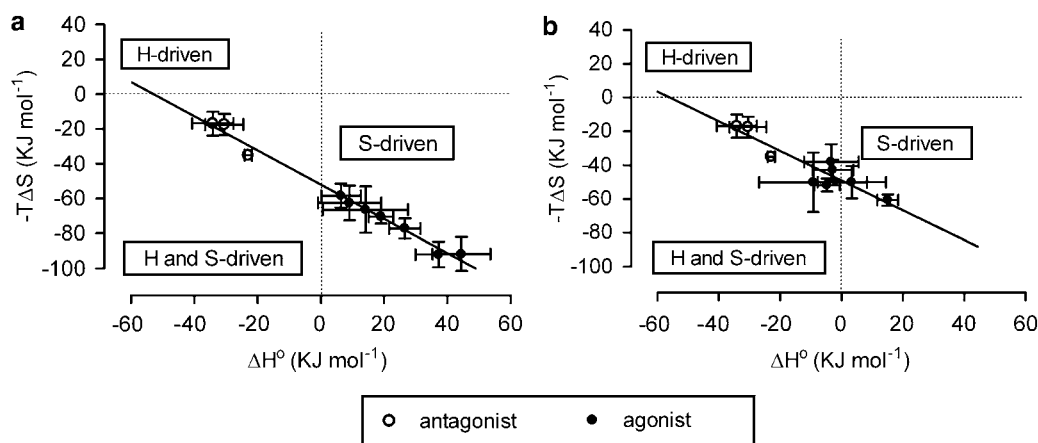
Ligand	$\Delta G^{or}$ calc ( $\text{kJ mol}^{-1}$ ) <sup>a</sup>	$\Delta G^{or}$ ( $\text{kJ mol}^{-1}$ )	$\Delta H^{or}$ ( $\text{kJ mol}^{-1}$ )	$-T\Delta S^{or}$ ( $\text{kJ mol}^{-1}$ )	$\Delta S^{or}$ ( $\text{J mol}^{-1} \text{K}^{-1}$ )	n
<i>Histamine H<sub>3</sub>-receptor agonists</i>						
imnepip	$-45.6 \pm 0.6$	$-45.6 \pm 0.7$	$15.1 \pm 3.4$	$-60.7 \pm 3.3$	$205.7 \pm 11.3$	5
imetit	$-44.8 \pm 1.0$	$-44.7 \pm 0.8$	$3.4 \pm 11.1$	$-50.0 \pm 9.6$	$169.6 \pm 32.6$	6
R- $\alpha$ -MH	$-39.2 \pm 2.2$	$-40.9 \pm 0.7$	$-9.2 \pm 17.6$	$-50.2 \pm 17.6$	$170.0 \pm 59.9$	5
chloroproxyfan	$-45.1 \pm 1.5$	$-43.7 \pm 0.7$	$-3.3 \pm 8.9$	$-38.1 \pm 10.4$	$136.8 \pm 28.7$	4
bromoproxyfan	$-45.9 \pm 1.4$	$-45.7 \pm 1.5$	$-2.8 \pm 6.6$	$-42.9 \pm 6.1$	$145.5 \pm 20.8$	4
iodoproxyfan	$-47.1 \pm 1.5$	$-47.2 \pm 1.6$	$-4.7 \pm 4.2$	$-51.8 \pm 3.9$	$175.5 \pm 13.1$	4
<i>Histamine H<sub>3</sub>-receptor antagonists</i>						
JB96132	$-50.6 \pm 0.8$	$-50.1 \pm 0.8$	$-20.9 \pm 9.4$	$-29.3 \pm 8.8$	$99.1 \pm 29.9$	4
thioperamide	$-50.1 \pm 0.8$	$-51.3 \pm 0.8$	$-40.7 \pm 6.4$	$-10.3 \pm 6.8$	$137.9 \pm 21.8$	5
[ <sup>3</sup> H]clobenpropit	$-54.4 \pm 1.5$	$-54.6 \pm 1.1$	$-10.5 \pm 8.5$	$-43.8 \pm 9.7$	$149.0 \pm 32.9$	3

Data are the mean  $\pm$  s.e.mean.

<sup>a</sup> $\Delta G^{or}$  at 21°C, calculated using equation 2.

In contrast, the extrathermodynamic plot indicated that the agonist binding was driven by an increase in entropy. There was a significant linear relationship between the two parameters ( $r=0.99$ ,  $P<0.01$ , slope =  $0.98 \pm 0.03$ , y-intercept =  $-51.9$ , x-intercept =  $-52.8$ ; Figure 10a).

The extrathermodynamic plot of  $\Delta H^{or}$  versus  $-T\Delta S^{or}$  for all ligands, obtained in buffer  $A_{Ca}$ , indicated that the binding of all ligands, except imetit and imnepip, was enthalpy- and entropy-driven (Figure 10b). There was a significant linear relationship between  $\Delta H^{or}$  and  $-T\Delta S^{or}$  ( $r=0.99$ ,  $P<0.01$ ,



**Figure 10** An extrathermodynamic plot for agonist and antagonist ligands in (a) buffer A and (b) buffer A<sub>Ca</sub>. The lines shown superimposed through the data were obtained by linear regression.

slope =  $0.88 \pm 0.13$ ,  $y$ -intercept =  $-53.9$ ,  $x$ -intercept =  $49.5$ ; Figure 10b).

## Discussion

In this study, we have determined the thermodynamic parameters underlying the binding of 10 ligands (seven agonists and three antagonists) at histamine H<sub>3</sub>-receptors of guinea-pig cortex. In previous studies, we have shown that agonist affinity values are overestimated in H<sub>3</sub>-receptor radioligand-binding assays, when buffer does not contain salts, and that the degree of affinity overestimation is correlated with the ligand's intrinsic activity ( $\alpha$ ) (see Harper *et al.*, 2007). In addition, we have shown that when H<sub>3</sub>-receptor radioligand-binding assays are conducted in the presence of salts,  $pK_i$  estimates are closer to  $pK_{app}$  values estimated by the method of Furchgott in the guinea-pig ileum bioassay (see Harper *et al.*, 2007). Therefore, to establish whether the thermodynamic parameters underlying the binding of agonists is dependent on whether agonist  $pK_i$  values are equivalent to  $pK_{app}$  values, we have determined agonist  $pK_i$  values in the presence and absence of buffer salts (buffer A<sub>Ca</sub> and buffer A).

Both kinetic studies and saturation studies of [<sup>3</sup>H]clobenpropit binding were performed at each assay temperature, in order to satisfy criteria which should be met when performing thermodynamic analysis of ligand binding, that is, that the binding should be to a homogeneous receptor population and that the binding of radioligand and ligand should reach equilibrium at each temperature. Saturation analysis was also performed at each temperature so that  $pIC_{50}$  values obtained in competition assays could be corrected for by any change in the occupancy of [<sup>3</sup>H]clobenpropit resulting from temperature-dependence of the  $pK_L$ . Saturation analysis confirmed that [<sup>3</sup>H]clobenpropit labelled a homogeneous population of H<sub>3</sub>-receptors in guinea-pig cerebral cortex in both the presence (buffer A<sub>Ca</sub>) and absence of buffer salt (300 mM CaCl<sub>2</sub>) (buffer A; Figures 2 and 3). Thus, at all incubation temperatures, the mean  $n_H$  parameter estimate was not significantly different from unity and the

H<sub>3</sub>-receptor density ( $B_{max}$ ) was not significantly changed (Figure 2, 3 and Table 1). Kinetic studies confirmed that the incubation time used for the saturation analysis (4 and 12°C = 24 h; 21 and 30°C = 2.75 h; Figure 5) was sufficient for equilibrium binding of the radioligand to have been achieved (3, 25, 30 and 80 min at 30, 21, 12 and 4°C) so that the  $pK_L$  values estimated at each temperature were correct. The dissociation rate ( $t_{1/2}$ ) of [<sup>3</sup>H]clobenpropit, determined in the kinetic studies at each temperature, indicated that the incubation times of 2.75 h at 21 and 30°C and of 24 h at 4 and 12°C, used for competition studies, were sufficient for equilibrium binding of agonist and antagonist ligands and thus for the accurate determination of  $pIC_{50}$  values. This was because in each case the incubation time was in excess of five times the  $t_{1/2}$  (Figure 6; see Motulsky and Mahan, 1983).

In the competition studies performed in buffer A, the finding that mean  $n_H$  parameter estimates for some of the agonist ligands at 21°C were significantly less than unity was consistent with our previous studies (Harper *et al.*, 2007). This agonist behaviour at 21°C and also that detected for some agonists at 4, 12 and 30°C, can be explained by the extended ternary complex or cubic ternary complex model (TCM) (Samama *et al.*, 1993; Lefkowitz *et al.*, 1993; Weiss *et al.*, 1996). In the extended TCM, it is proposed that the receptor can exist in a low-affinity state (R) and a high-affinity state (R\*) which can also interact with a G-protein (R\*G) in the absence of the agonist. Thus, flat competition curves are observed because the agonist binds with high affinity to R\* or R\*G and with lower affinity to R. In the cubic TCM, it is postulated that the receptor can exist in a low-affinity state (R<sub>i</sub> equivalent to R in the TCM) and a high-affinity state (R<sub>a</sub>, equivalent to R\* in the TCM) and that both R and R\* can exist as high-affinity states as a consequence of interaction with a G-protein (RG and R\*G). Therefore, according to this model, it is possible to obtain flat competition curves because the agonist binds with low affinity to R and with high affinity to preformed high-affinity receptor states (R\* and R\*G). However, flat competition curves could also arise because the agonist induces varying amounts of receptor to form a

high-affinity state (AR\*G or ARG) through binding to low-affinity receptors (R).

The finding of mean  $n_H$  values significantly less than unity for the antagonist thioperamide at 4 and 21°C in buffer A and at 4°C in buffer A<sub>Ca</sub>, may be a result of type 1 error. This is because at 21°C, at least, this contrasts with our previous studies where we found that  $n_H$  was not different from unity (Harper *et al.*, 2007). In addition, we found that there was no significant difference between the fit of individual replicate thioperamide competition curves, obtained at 4 and 21°C, to the Hill equation and to the Hill equation with  $n_H$  constrained to unity, as judged by an *F*-test. Type 1 error could also explain the finding of  $n_H$  values significantly less than unity for iodoproxyfan and *R*- $\alpha$ -MH at 12°C and imipip at 4 and 12°C, under assay conditions (buffer-containing salts, buffer A<sub>Ca</sub>) which we have previously suggested, provide a measure of the agonist  $pK_{app}$  (Harper *et al.*, 2007) and therefore under conditions where we would expect  $n_H$  not to be different from unity. In support of this, when the mean agonist  $n_H$  values were significantly less than unity, there was no significant difference between the fit of the individual replicate competition curves to the Hill equation and to the Hill equation with  $n_H$  constrained to unity, as judged by an *F*-test.

In light of the finding of  $n_H$  values which were significantly less than unity for some ligands at some of the temperatures, it could be argued that we should not have derived  $pK_i$  values from  $pIC_{50}$  values using the Cheng–Prusoff equation. This is because the derivation of this correction relies on simple competition between two ligands at a homogenous receptor population and, therefore, should be applied only when  $n_H$  is not different from unity. However, in this study, we have corrected all  $pIC_{50}$  values using the Cheng–Prusoff equation to correct for the differential occupancy of ~0.2 nM [<sup>3</sup>H]clobenpropit in the two buffers (buffer A and buffer A<sub>Ca</sub>) at the different incubation temperatures. Ideally, for this not to be a confounding problem, we would have performed competition studies in both buffers at each temperature at [<sup>3</sup>H]clobenpropit concentrations equivalent to the  $K_L$ . However, this was not possible because the low specific activity of the radioligand resulted in too small a specific binding window in buffer A at the ligand concentration ideally required for studies at 4°C (~0.03 nM).

It could also be reasoned that the data could have been analysed using a two-site model and then the  $pIC_{50}$  values corrected using the Cheng–Prusoff equation to provide  $pK_H$  and  $pK_L$  parameters. However, we chose not to analyse the data in this way because it would not have been possible to obtain these parameters for all the agonists at all the temperatures. This is because for some agonists at some temperatures, the mean  $n_H$  value was not different from unity and for individual replicate competition data, the fit to the Hill equation was not significantly different to the fit to the Hill equation with  $n_H$  constrained to unity (e.g. proxyfan at 4, 12, 21 and 30°C; bromoproxyfan at 4 and 21°C, Table 3).

It is noteworthy that the affinity values obtained for all the ligands in buffer A at 21°C were consistent with observations made in a previous study using the same assay buffer and incubation temperature, as was the  $pK_i$  value for *R*- $\alpha$ -MH at

21°C in buffer A<sub>Ca</sub> (Harper *et al.*, 2007). In addition, it is notable that the  $pK_i$  values for all the agonists, at 21°C in buffer A<sub>Ca</sub>, were associated with  $n_H$  values that were not different from unity and, moreover, that were not different from agonist  $pK_{app}$  values estimated previously in the guinea-pig ileum (see Harper *et al.*, 2007).

The discovery of thermodynamic discrimination of agonists and antagonists, at the H<sub>3</sub>-receptor (Figure 10a, antagonist binding =  $\Delta H^\circ$ - and  $\Delta S^\circ$ -driven; agonist binding =  $\Delta S^\circ$ -driven) mirrors that reported for binding of agonist and antagonist ligands to the adenosine A<sub>1</sub> receptor (Dalpiaz *et al.*, 2000), adenosine A<sub>2A</sub> receptor (Borea *et al.*, 1996b), GABA<sub>A</sub> receptor (Maksay, 1994) and serotonin 5-HT<sub>3</sub> receptor (Borea *et al.*, 1996a). The finding that the binding of all the H<sub>3</sub>-receptor ligands, investigated in this study was, at least in part, entropy driven could be explained by disorganisation of a solvation sphere around the ligands as they bind to the receptor; this is because all the ligands contain an imidazole moiety and this would be protonated at the assay pH of 7.4. It would have been interesting to determine whether the binding of some of the recently described non-imidazole histamine H<sub>3</sub>-receptor antagonists (e.g. Linney *et al.*, 2000; Lazewska *et al.*, 2002; Meier *et al.*, 2002; Shah *et al.*, 2002; Apodaca *et al.*, 2003; Chai *et al.*, 2003; Miko *et al.*, 2003, 2004; Turner *et al.*, 2003; Zaragoza *et al.*, 2004, 2005; Sun *et al.*, 2005; Lazewska *et al.*, 2006; Rivara *et al.*, 2006) are also, in part, entropy driven.

It has been suggested for the ligand-gated ion channels, where the thermodynamic discrimination of agonists and antagonists mirrors that of the H<sub>3</sub>-receptor, that this phenomenon can be explained by both interaction of the ligand with the receptor and variation in the water-accessible surface which occurs when the agonist induces channel opening (Borea *et al.*, 1998). Therefore, a possible explanation for the thermodynamic discrimination of H<sub>3</sub>-receptor agonists and antagonists is that the agonists induce a change in receptor conformation, perhaps into a less-constrained state, which in turn, leads to the formation of a ternary complex with a G-protein and this consequently results in a decrease in the solvation of the cytosolic side of the receptor. The finding of a decrease in enthalpy associated with antagonist binding may be explained by hydrogen bond formation and van der Waals interactions occurring between the ligands and the binding pocket which cannot be compensated for by changes in entropy that result from agonist-induced conformational changes in the receptor.

The hypothesis that agonist binding at H<sub>3</sub>-receptors in guinea-pig cerebral cortex induces ternary complex formation (ARG or AR\*G) and this brings about the large increase in entropy, is supported by the finding that under conditions (buffer A<sub>Ca</sub>) where the agonists express  $pK_i$  values that are closely correlated with their  $pK_{app}$  values (see Harper *et al.*, 2007), agonist and antagonists cannot be discriminated thermodynamically (Figure 10b). Thus, in buffer A<sub>Ca</sub>, the agonist  $pK_i$  values remained unchanged with decreasing temperature in the same manner as the antagonist ligands (Table 3). In addition, van't Hoff plots constructed from agonist thermodynamic data obtained in buffer A<sub>Ca</sub> (Figure 9), are similar to those of the antagonists [<sup>3</sup>H]clobenpropit (Figure 5), thioperamide and JB96132 (Figure 8).

Despite the gross thermodynamic discrimination between the agonist and antagonist ligands at H<sub>3</sub>-receptors (Figure 10a), it was surprising that there was no relationship between agonist intrinsic activity ( $\alpha$ ), measured previously in a guinea-pig ileum bioassay (see Harper *et al.*, 2007) and either mean  $\Delta H^{\circ}$  or mean  $\Delta S^{\circ}$  (see Table 4; e.g. *R*- $\alpha$ -MH,  $\Delta H^{\circ} = 19.1$ ,  $\Delta S^{\circ} = 218.3$ ,  $\alpha = 1.0$ ; proxyfan,  $\Delta H^{\circ} = 44.5$ ,  $\Delta S^{\circ} = 311.3$ ,  $\alpha = 0.35$ ) even for the analogues of proxyfan which differ only in the halogen substitution at the *para* position of the benzene ring (Figure 1) (Table 4; proxyfan  $\Delta S^{\circ} = 311.3$ ,  $\Delta H^{\circ} = 44.5$ ,  $\alpha = 0.35$ ; iodoproxyfan  $\Delta S^{\circ} = 311.0$ ,  $\Delta H^{\circ} = 37.3$ ,  $\alpha = 0.90$ ). Failure to find a relationship between intrinsic activity and thermodynamic parameters may have been a consequence of the experimental design. This is because we have previously found a significant effect of tissue preparation on the agonist  $pK_i'$  values under conditions where this parameter is not equivalent to the  $pK_{app}$  (see Harper *et al.*, 2007). Ideally, we would have generated competition curve data for all the ligands at each temperature on the same experimental day, however, this was not possible due to restrictions in the availability of equipment required to accurately maintain temperatures of 4, 12, 21 and 30°C.

It is perplexing that the entropy associated with the binding of antagonists is increased in the presence of buffer salts (buffer A<sub>Ca</sub>) despite an overall decrease in the entropy of agonist binding (Tables 4 and 5; e.g. clobenpropit buffer A  $\Delta S^{\circ} = 119.5$ ; buffer A<sub>Ca</sub>,  $\Delta S^{\circ} = 149.0$ ; iodoproxyfan buffer A  $\Delta S^{\circ} = 311.0$ , buffer A<sub>Ca</sub>,  $\Delta S^{\circ} = 175.5$ ). This may be a consequence of the buffer salts increasing the hydration of the ligands so that it is necessary for more water to be stripped away upon receptor binding. This possibility may explain why although the overall entropy associated with agonist binding is reduced in the presence of salts and in conditions where the agonist affinity is similar to  $pK_{app}$ , it is not as low as that of antagonist binding in the absence of buffer salts (buffer A). It would be interesting to repeat these thermodynamic studies to establish whether agonist  $\Delta S^{\circ}$  values are lower when G-protein coupling of the H<sub>3</sub>-receptor is prevented, perhaps by *Pertussis* toxin treatment, but where there is not the potential complication of changes in ligand hydration.

The linear relationship between mean  $\Delta H^{\circ}$  and  $T\Delta S^{\circ}$  for the H<sub>3</sub>-receptor was not unexpected and has been reported for numerous G-protein-coupled receptors (GPCRs) and ligand-gated ion channels (e.g.  $\beta$ -adrenoceptor, adenosine A<sub>1</sub>, adenosine A<sub>2A</sub>, dopamine D<sub>2</sub>, serotonin 5-HT<sub>1A</sub>, glycine, GABA<sub>A</sub>, and nicotinic receptor; see Borea *et al.*, 1998). The linear relationship indicates that enthalpy–entropy compensation exists for the H<sub>3</sub>-receptor, that is, changes in enthalpy are compensated for by changes in entropy (or vice versa) such that the free-energy change  $\Delta G^{\circ}$  is constant.

## Conclusion

The binding of agonists and antagonists at the histamine H<sub>3</sub>-receptor can be thermodynamically discriminated when agonist  $pK_i$  values are not equivalent to  $pK_{app}$  values; agonist binding is entropy-driven and antagonist binding enthalpy- and entropy-driven. In the presence of buffer salts, where the ligand  $pK_i$  values are more closely correlated with their  $pK_{app}$

values estimated in a functional bioassay, the thermodynamic parameters underlying agonist binding are changed and are not different from those of antagonists.

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## Conflict of interest

The authors state no conflict of interest.

## References

- Alves-Rodrigues A, Lemstra S, Vollinga RC, Menge WMPB, Timmerman H, Leurs R (2001). Pharmacological analysis of imipip and imetit homologues. Further evidence for histamine H<sub>3</sub> receptor heterogeneity. *Behav Brain Res* **124**: 121–127.
- Apodaca R, Dvorak CA, Xiao W, Barbier AJ, Boggs JD, Wilson SJ *et al.* (2003). A new class of diamine-based human histamine H<sub>3</sub> receptor antagonists: 4-(aminoalkoxy)benzylamines. *J Med Chem* **46**: 3938–3944.
- Aronstam RS, Narayanan TK (1988). Temperature effect on the detection of muscarinic receptor-G protein interactions in ligand binding assays. *Biochem Pharmacol* **37**: 1045–1049.
- Borea PA, Dalpiaz A, Gessi S, Gilli G (1996a). Thermodynamics of 5-HT<sub>3</sub> receptor binding discriminates agonistic from antagonistic behaviour. *Eur J Pharmacol* **298**: 329–334.
- Borea PA, Dalpiaz A, Varani K, Gessi S, Gilli G (1996b). Binding thermodynamics at A<sub>1</sub> and A<sub>2A</sub> adenosine receptors. *Life Sci* **59**: 1373–1388.
- Borea PA, Varani K, Gessi S, Gilli P, Gilli G (1998). Binding thermodynamics at the human neuronal nicotine receptor. *Biochem Pharmacol* **55**: 1189–1197.
- Chai W, Breitenbucher JG, Kwok A, Li X, Wong V, Carruthers NI *et al.* (2003). Non-imidazole heterocyclic histamine H<sub>3</sub>-receptor antagonists. *Bioorgan Med Chem Lett* **13**: 1767–1770.
- Cheng YC, Prusoff WH (1973). Relationship between the inhibition constant  $K_i$  and the concentration of inhibitor which causes 50% inhibition  $IC_{50}$  of an enzymic reaction. *Biochem Pharmacol* **22**: 3099–3108.
- Dalpiaz A, Borea PA, Gessi S, Gilli G (1996). Binding thermodynamics of 5-HT<sub>1A</sub> receptor ligands. *Eur J Pharmacol* **312**: 107–114.
- Dalpiaz A, Scatturin A, Varani K, Pecoraro R, Pavan B, Borea PA (2000). Binding thermodynamics and intrinsic activity of adenosine A<sub>1</sub> receptor ligands. *Life Sci* **67**: 1517–1524.
- Duarte EP, Oliveira CR, Caravvalho A (1988). Thermodynamic analysis of antagonist and agonist interactions with dopamine receptors. *Eur J Pharmacol* **147**: 227–239.
- Harper EA, Gardner B, Shankley NP, Black JW (2002). Histamine H<sub>3</sub>-receptor agonists and antagonists can be thermodynamically-discriminated. *Br J Pharmacol* **186P**.
- Harper EA, Shankley NP, Black JW (1999a). Evidence that histamine homologues discriminate between H<sub>3</sub>-receptors in guinea-pig cerebral cortex and ileum longitudinal muscle myenteric plexus. *Br J Pharmacol* **128**: 751–759.
- Harper EA, Shankley NP, Black JW (1999b). Characterisation of the binding of [<sup>3</sup>H]-clobenpropit to histamine H<sub>3</sub>-receptors in guinea-pig cerebral cortex membranes. *Br J Pharmacol* **128**: 881–890.
- Harper EA, Shankley NP, Black JW (2007). Correlation of apparent affinity values obtained in a H<sub>3</sub>-receptor binding assay with apparent affinity ( $pK_{app}$ ) and intrinsic activity ( $\alpha$ ) estimated in functional bioassay. *Br J Pharmacol* [E-pub ahead of print: 12 March 2007; doi:10.1038/sj.bjp.0707174].
- Hitzeman R (1988). Thermodynamic aspects of drug–receptor interactions. *Trends Pharmacol Sci* **9**: 408.

- Kilpatrick GJ, Tayar NEL, Van De Waterbeemb H, Testa JB, Marsden CD (1986). The thermodynamics of agonist and antagonist binding to dopamine D-2 receptors. *Mol Pharmacol* **30**: 226–234.
- Lazewska D, Kiec-Kononowicz K, Pertz HH, Elz S, Stark H, Schunack W (2002). Piperidine-containing histamine H<sub>3</sub> receptor antagonists of the carbamate series: the influence of the additional ether functionality. *Pharmazie* **57**: 791–795.
- Lazewska D, Ligneau X, Schwartz J-C, Schunack W, Stark H, Kiec-Kononowicz K (2006). Ether derivatives of 3-piperidinopropanol-1-ol as non-imidazole histamine H<sub>3</sub> receptor antagonists. *Bioorg Med Chem Lett* **14**: 3522–3529.
- Lefkowitz RJ, Cotecchia S, Samama P, Costa T (1993). Constitutive activity of receptors coupled to guanine nucleotide regulatory proteins. *Trends Pharmacol Sci* **14**: 303–307.
- Li J-G, Raffa RB, Cheung P, Tzeng T-B, Liu-Chen L-Y (1998). Apparent thermodynamic parameters of ligand binding to the cloned rat  $\mu$ -opioid receptor. *Eur J Pharmacol* **354**: 227–237.
- Linney ID, Buck IM, Harper EA, Kalindjian SB, Pether MJ, Shankley NP *et al.* (2000). Design, synthesis and structure–activity relationships of novel non-imidazole histamine H<sub>3</sub> receptor antagonists. *J Med Chem* **43**: 2362–2370.
- Maguire PA, Loew GH (1996). Thermodynamics of ligand binding to the cloned  $\delta$ -opioid receptor. *Eur J Pharmacol* **318**: 505–509.
- Maksay G (1994). Thermodynamics of  $\gamma$ -aminobutyric acid type A receptor binding differentiate agonists from antagonists. *Mol Pharmacol* **46**: 386–390.
- Meier G, Ligneau X, Pertz HH, Ganellin CR, Schwartz J-C, Schunack W *et al.* (2002). Piperidino-hydrocarbon compounds as novel non-imidazole histamine H<sub>3</sub>-receptor antagonists. *Bioorg Med Chem Lett* **10**: 2535–2542.
- Miko T, Ligneau X, Pertz HH, Arrang J-M, Ganellin CR, Schwartz J-C *et al.* (2004). Structural variations of 1-(4-(phenoxymethyl)benzyl) piperidines as non-imidazole histamine H<sub>3</sub> receptor antagonists. *Bioorg Med Chem* **12**: 2727–2736.
- Miko T, Ligneau X, Pertz HH, Ganellin R, Arrang J-M, Schwartz J-C *et al.* (2003). Novel non-imidazole histamine H<sub>3</sub>-receptor antagonists: 1-(4-(phenoxymethyl)benzyl)piperidines and related compounds. *J Med Chem* **46**: 1523–1530.
- Mohler H, Richards JG (1981). Agonist and antagonist benzodiazepine receptor interaction *in vitro*. *Nature* **24**: 763–765.
- Motulsky HJ, Mahan LC (1983). The kinetics of competitive radioligand binding predicted by the law of mass action. *Mol Pharmacol* **25**: 1–9.
- Raffa RB, Porreca F (1989). Thermodynamic analysis of the drug–receptor interactions. *Life Sci* **44**: 245.
- Reith MAE, Sershen H, Lajtha A (1984). Thermodynamics of the interaction of tricyclic drugs with binding sites for <sup>3</sup>H-imipramine in mouse cerebral cortex. *Biochem Pharmacol* **33**: 4101–4104.
- Rivara M, Zuliani V, Cocconelli G, Morini G, Comini M, Rivara S *et al.* (2006). Synthesis and biological evaluation of new non-imidazole H<sub>3</sub>-receptor antagonists of the 2-aminobenzimidazole series. *Bioorg Med Chem Lett* **14**: 1413–1424.
- Samama P, Cotecchia S, Costa T, Lefkowitz RJ (1993). A mutation-induced activated state of the  $\beta_2$ -adrenergic receptor. Extending the ternary complex model. *J Biol Chem* **268**: 4625–4636.
- Shah C, McAtee L, Breitenbucher JG, Rudolph D, Li X, Lovenberg TW *et al.* (2002). Novel human histamine H<sub>3</sub> receptor antagonists. *Bioorg Med Chem Lett* **12**: 3309–3312.
- Sun M, Zhao C, Gfesser GA, Thifault C, Miller TR, March K *et al.* (2005). Synthesis and SAR of 5-amino- and 5-(aminoethyl)benzofuran histamine H<sub>3</sub> receptor antagonists with improved potency. *J Med Chem* **48**: 6482–6490.
- Testa B, Jenner P, Kilpatrick GJ, Eltayar N, Van De Waterbeemb H, Marsden CD (1987). Do thermodynamic studies provide information about both binding to and the activation of dopaminergic and other receptors? *Biochem Pharmacol* **36**: 4041–4046.
- Todd RD, Babiniski J (1987). A thermodynamic study of 5-[<sup>3</sup>H]hydroxytryptamine binding to human cortex membranes. *J Neurochem* **49**: 1480–1483.
- Turner SC, Esbenshade TA, Bennani YL, Hancock AA (2003). A new class of histamine H<sub>3</sub>-receptor antagonists: synthesis and structure-activity relationships of 7,8,9,10-tetrahydro-6H-cyclohepta[b]-quinolines. *Bioorg Med Chem Lett* **12**: 2131–2135.
- Weiland GA, Minneman KP, Molinoff PB (1979). Fundamental difference between the molecular interactions of agonists and antagonists with the  $\beta$ -adrenergic receptor. *Nature* **281**: 114.
- Weiss JM, Morgan PH, Lutz MW, Kenakin TP (1996). The cubic ternary complex receptor–occupancy model I. Model Description. *J Theor Biol* **178**: 151–167.
- Zahniser NR, Molinoff PB (1983). Thermodynamic differences between agonist and antagonist interaction with binding sites for <sup>3</sup>H-spiroperidol in rat striatum. *Mol Pharmacol* **23**: 303.
- Zaragoza F, Stephensen H, Knudsen SM, Pridal L, Wulff BS, Rimvall K (2004). 1-alkyl-4-acylpiperazines as a new class of imidazole-free histamine H<sub>3</sub> receptor antagonists. *J Med Chem* **47**: 2822–2838.
- Zaragoza F, Stephensen H, Peschke B, Rimvall K (2005). 2-(4-alkylpiperazin-1-yl)quinolines as a new class of imidazole-free histamine H<sub>3</sub>-receptor antagonists. *J Med Chem* **48**: 306–311.